system at different temperatures.(NB Teut=T4) Based on this data: 1. Figure 1 shows Gibbs energy - composition diagrams from hypothetical eutectic A-B

Construct the binary phase diagram. (3p)

value. (1p) by Estimate the composition of the eutectic point. Explain shortly, how you obtained the

 δ Show graphically what is the chemical potential of A and B in α - and β - phases (i.e.

 μ_A^{α} , μ_B^{β} , μ_B^{α} , μ_B^{β}), when T=T₅ and x_B =0.5. (1p) NB! You can used the attached template to give your answers, add your name and student number.

shown in Fig. 2: 2. Co-based alloys are commonly used for implants. Based on the Co-Mo phase diagram

What is the solubility of Molybdenum in αCobalt and Cobalt in Molybdenum at 1200°C? (1p)

 \forall) If the nominal composition of the CoMo alloy is $x_{Co}=0.5$ and $T=800^{\circ}$ C, what phases are in equilibrium and what are their compositions?(1p)

amounts of phases as a function of temperature) from 2800°C to 400°C when the nominal W Please explain the equilibrium (D_{Liq}=D_{solid}=∞) solidification (i.e. the compositions and composition of the alloy is $x_{Mo}=0.92$. (2p)

by If microsegregation ($D_{Liq}=\infty$ and $D_{solid}=0$) occurs, how would the solidification of the same alloy $[x_{Mo}=0.92]$ change? (1p)

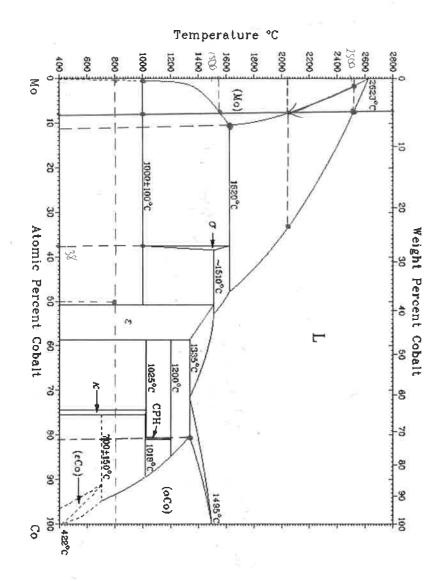


Figure 2 Co-Mo binary phase diagram

changed if the temperature is then decreased down to 650°C? (4+1p) achieved in all interfaces. (NB. Mark clearly the compositions of the phases at all for both components, when sufficient amount of time has elapsed so that local equilibria is diagram (see Fig. 3) draw the structure of the diffusion couple and the composition profiles interfaces!) In addition, please explain shortly how the structure of the diffusion couple is =semi-infinite diffusion couple) at the temperature of 1000°C. Based on the phase Thick Cobalt (Co) and Titanium (Ti) metal plates are attached against each other

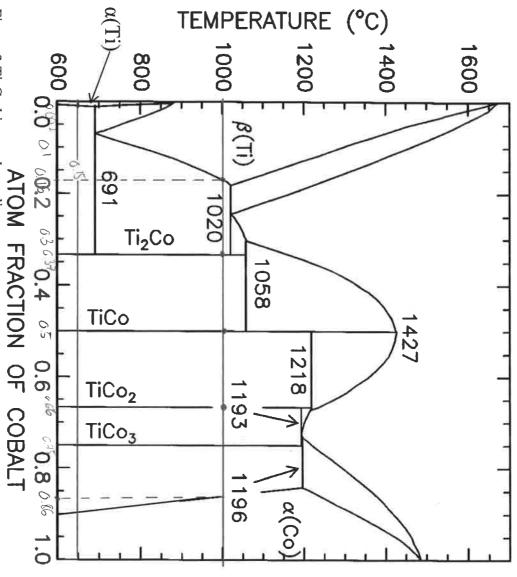


Figure 3 Ti-Co binary phase diagram