

PHYS-C0256 - Thermodynamics and Statistical Physics
Exam on February 22, 2023

7 problems - 30 points

1. (Concepts, 3 points) Write simple explanation for each of the following
- Thermodynamic limit
 - Clausius inequality
 - Third law of thermodynamics

2. (Probabilities, 3 points) There are 10 freely moving non-interacting particles in a box with volume V , see Fig. 1. A wall is inserted such that it divides the volume of the box into two equal halves (volumes $V/2$), say "left" and "right" half. Find the probability that after insertion of the wall
- there are no particles in the "left" half.
 - only one particle is found in the "left" half.
 - 5 particles are found on each side.

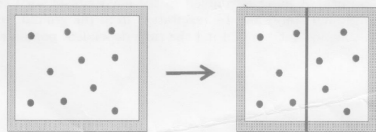


FIG. 1. Problem 2.

3. (Canonical distribution, partition function, 4 points) A three level system (see Fig. 2) has energy level separation $\Delta E = E_e - E_g$ between the first excited state (e) and the ground state (g). There is the third level (f) for which $E_f - E_e = 0.9\Delta E$. The system is canonical as it is coupled to a heat bath at temperature T . Find the equilibrium occupation probabilities p_f, p_e, p_g in each state. Estimate how low should the temperature ($k_B T / \Delta E$) be to have $p_e < 10^{-3}$. [$\ln(10) = 2.3 \dots$]

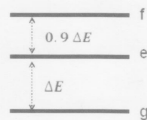


FIG. 2. Problem 3.

4. (Ideal gas, 4 points) Find the ratio of the isothermal and adiabatic compressibilities κ_T / κ_S .

5. **(Heat, adiabatic process, entropy, 6 points)** Consider one mole of van der Waals gas with equation of state $p = \frac{RT}{V-b} - \frac{a}{V^2}$, where a , b and R are constants.

- (a) Calculate the amount of heat produced in a reversible isothermal expansion from V_1 to V_2 .
 (b) Show that in an adiabatic process $T(V-b)^{R/C_V} = \text{const.}$, where c_v is heat capacity for one mole of gas at constant volume.
 (c) Calculate the entropy of this gas.

6. (a) **(Normal metal-Insulator-Superconductor NIS junction, 5 points)** For this junction at low temperature T , $eV, k_B T \ll \Delta$, starting from the equation $I = \frac{1}{eR_T} \int_{-\infty}^{\infty} de n_S(e) [f_N(\epsilon - eV) - f_S(\epsilon)]$, show that the following equation holds approximately for its current I versus voltage V dependence,

$$I = I_0 e^{-(\Delta - eV)/k_B T}, \quad (1)$$

where $I_0 = \frac{1}{eR_T} \sqrt{\frac{\pi \Delta k_B T}{2}}$, Δ the superconducting gap, and $n_S(\epsilon)$ is the density of states of the superconductor. Show further that $\frac{d}{dV}(\ln I)$ gives temperature dictated just by constants of nature and temperature.

(b) **(Efficiency of a refrigerator, 2 points)** The cooling power of N in the same configuration is maximized at $V \simeq \Delta/e$. A more precise calculation shows that the current at this point is $I = 0.48 \frac{\Delta}{eR_T} \sqrt{k_B T/\Delta}$ and the cooling power has value $\dot{Q}_{\text{NIS}} = 0.59 \frac{\Delta^2}{e^2 R_T} (k_B T/\Delta)^{3/2}$. Write the efficiency η of the NIS refrigerator at this optimum point.

7. **(Qubit relaxation, 3 points)** A qubit is coupled to the environment such that it has transition rates Γ_1 (relaxation from the excited to ground state) and Γ_+ (excitation from the ground to excited state). The qubit is initiated in the excited state at time instant $t = 0$. Find the time dependent population of the qubit in the ground state after the initialization.