## Midterm Exam 1 23.3.2011

- 1. Define or explain the following as accurately and concisely as you can:
  - (a) order parameter
  - (b) phase space
  - (c) phase of matter
  - (d) thermodynamic limit
  - (e) Liouville equation
  - (f) phase transition
- 2. Use the microcanonical formalism to calculate the entropy S(E,V,N) of a classical ideal gas.

Hint:

$$V_N(R) = \frac{\pi^{N/2}}{(N/2)\Gamma(N/2)}R^N$$

- 3. Consider an ideal gas of massless photons, with spin S=1 and energy  $E_i=\hbar\omega_i=\hbar c k_i$ . Photons have two polarized spin states, i.e. degeneracy g=2.
  - (a) The number of photons in a closed system is not conserved. Use the canonical ensemble to show that as a consequence, the chemical potential of the gas is zero.
  - (b) Show by using the grand canonical ensemble that the average occupation number of photons at state  $\omega$  is given by

$$n(\omega) = \frac{2}{e^{\beta\hbar\omega} - 1}$$

4. Consider an interacting (monoatomic) gas, whose Hamiltonian is given by

$$H = \sum_{i=1}^{N} \left[ \frac{\vec{p}_i^2}{2m} + \sum_{j=1}^{i-1} V(r_{ij}) \right],$$

where  $r_{ij} = |\vec{r}_i - \vec{r}_j|$  is the distance between atoms i and j, and  $V(r_{ij})$  is their pair interaction potential.

(a) Show that the canonical partition function can be written as

$$Z_N(T,V) = \frac{V^N}{N! \lambda_T^{3N}} Q_N(T,V),$$

where  $\lambda_T = \frac{h}{\sqrt{2\pi m k_B T}}$  is the thermal de Broglie wavelength. What is  $Q_N(T,V)$ ?

(b) Define the Meyer functions  $f_{ij} = f(r_{ij}) = e^{-\beta V(r_{ij})} - 1$ . Show that with these

$$Q_N(T,V) = \frac{1}{V^N} \int d\vec{r}_1 \cdots \int d\vec{r}_N \prod_{i < j} (1 + f_{ij}).$$

(c) In the limit where  $f_{ij} \ll 1$ , show that  $Q_N$  can be approximated by

$$Q_N(T,V) \approx 1 + \frac{N(N-1)}{2V^2} \int d\vec{r_1} \int d\vec{r_2} f(|\vec{r_1} - \vec{r_2}|).$$

When is this a reasonable approximation? Hint: Consider the coefficient in front of the next order term.

Recall:  $\int_{-\infty}^{\infty} e^{-ax^2} dx = \sqrt{\frac{\pi}{a}}$